How to Cite:

Mohamned, M. N., & Jawad, M. H. (2022). Synthesis of new Schiff bases derivative of benzoin compounds. *International Journal of Health Sciences*, *6*(S4), 6115–6121. https://doi.org/10.53730/ijhs.v6nS4.9542

Synthesis of new Schiff bases derivative of benzoin compounds

Moayd. N. Mohamned

Department of Chemistry-College of Science-Qadisiyah University Corresponding author email: sci.chem.mas.20.27@qu.edu.iq

Mustafa. H. Jawad

Department of Chemistry-College of Science-Qadisiyah University Email: moayad.alshbani@qu.ed.id

Abstract---In this In this research study a different chain of schiff bases starting benzoin 4,4-dimethyl benzoin/4,4-dichloro benzoin/4,4-dibromo benzoin/4-chloro-4-dimethylamino benzoin/4-methyl benzoin with 1)Alanine yield series of new schiff bases, the structure of the compound was determined on the bases of their FT-IR,1HNMR and C.H.N. the pureness of the created compounds were tested via the antibacterial action were estimated by paper disc dispersal technique.

Keywords---Schiff bases, benzoin, alanine, biological activity.

Introduction

Schiff bases are described by the (C=N) (ktimine) cluster which is key for elucidating the appliance of transamination besides racemization responses in organic systems (1,2). the response is conventional and taking in height incomes. In over-all ketones respond more little by little than aldehydes and upper hotness and lengthier response times are frequently requisite (3,4,5).Schiff bases actual main in preparation fresh centers, these compound existing abundant diversity of organic action reaching from Antifungal and bacteriocidal.

Materials and Methods

Melting points of the created compounds were resolute via exposed tube and are uncorrected the pureness of the compound was tested via get into TLC plates via (benzene: methanol (8:2) solvent scheme the advanced chromatographic plates were pictured below UV. IR scales were noted via KBr on FTIR. C.H.N analyzer and H-NMR spectra (300 MHz) & so tasters at Al Albeit university.

International Journal of Health Sciences ISSN 2550-6978 E-ISSN 2550-696X © 2022.

Manuscript submitted: 27 March 2022, Manuscript revised: 9 May 2022, Accepted for publication: 18 June 2022

Synthesis of benzoin

In absolute ethanol (30ml)dissolved (0.01) mole p-methyl benzaldehyde) the reaction mixture was refluxed on water bath when hot mixture added (0.066 mole) KCN in (20 ml) water and refluxed 3 hours.Then it is left to cool 24 hours.The basic produce was cleaned by recrystallization of ethanol.

Creation of new Schiff bases

in 50 ml of absolute ethanol mix Benzoin derivatives (0.02 mole) with duobs & glacial acetic acid were stirred for (15 min), then (0.02 mole) alanine was additional and moved for (4h) The hard made was clarified and recrystallized from ethanol .



Synthesis of Schiff bases

Results and Discussions

The preparation of compounds Schiff bases can be make clear by the next suggested mechanical The recently prepared Schift bases described by spectral (U.V,IR,)(C.H. N) and H NMR. The physical description and spectral information are existing in tables 1,2,3. Spectra presented vanishing of (NH2) and vanishing of (C=O) absorption band in benzoin. absorption band in the other new derivative at (1550-1590) cm to (C=N) group. Schiff bases were verified beside for each of apositive and negative grm the outcomes of antibacterial being there in table 4.

	Table (1)	
Physical	data of the new Schiff bases	s

Com.	formula	Color.	Yield %,	(ca	alc-). foi	und%
1	C ₂₀ H ₂₃ N2O ₃₃	Brown	70	С	Η	Ν

6116

				70.77	6.83	8.25
				70.71	6.77	8.18
2	C. IL NO	0	73.29	6.80	4.50	
	$C_{19}\Pi_{21} NO_{3}$	Orange	60	73.22	6.73	4.43
3	C. U. CLNO	Vallow	60	57.97	4.29	3.98
	$C_{17}\Pi_{15}CI_{2}INO_{3}$	renow	09	57.91	4.23	3.91
4	C. H. CINOO	White	69	63.24	5.87	7.76
	$C_{19}\Pi_{21}CINZO_{3}$	WIIIte	08	63.17	5.80	7.69
5	CULUNO	H ₁₉ NO ₃ White 55	55	72.71	6.44	4.71
			55	72.64	6.38	4.64



Figure(1) FT-IR for compound 1



Figure (2) FT-IR for.compound 5

COM.	Other cm-1 v	C=C	C=N	ОН
1	C-N(CH ₃)2(1265)	1600	1630	3270-3500
2		1590	1620	3270-3500
3	C-CL(733)	1595	1610	3270-3500
4		1578	1646	3270-3500
5		1600	1672	3270-3500

Table.(2) FT-IR Data of Schiff base



Figure(3) 1HNMR for.compound 1



Figure(4) 1HNMR for.compound 5

Com	.Structure.	Chemical.shift
1	N H ₃ C COOH	(Ph)6.2-7.5 ppm (ph-N)3.5 ppm (OH) 9.3 ppm (COOH)0.2 ppm
2	HO HO H3 CH HO COOH	(CH)1.3-1.4 ppm (Ph)6.2-7.5 ppm (OH) 9.2 ppm (COOH)0.2 ppm
3	CI H ₃ C COOH	(Ph)6.2-7.5 ppm (OH) 9.2 ppm (COOH)0.2 ppm (C-CL)7.3-8.78 ppm
4	N H ₃ C COOH	(Ph)6.2-7.5 ppm (OH) 9.3ppm (COOH)0.2 ppm (C-CL)7.3-8.78 ppm (ph-N)3.4 ppm
5		(CH)1.3-1.4 ppm (Ph)6.2-7.5 ppm (OH) 9.1ppm (COOH)0.2 ppm

Table,(3) Data ,for new schiff bases

Table (4) Biological.activity

Bacteria(gram)	positive	negative
Comp.	S-aureus	E-coli
1	4 mm	7 mm
2	7 mm	8 mm
3	6 mm	7 mm
4	9 mm	10 mm
5	12 mm	10 mm

Conclusion

The effect of prepared compounds on two types of gram-negative and grampositive bacteria has been studied and these species have been tested in view of their importance in the medical field because they cause many and varied diseases as well as differ in their resistance to antibiotics. These bacteria were isolated from clinical cases after confirming their diagnosis based on biochemical characteristics.

Acknowledgments

I would like to thank the Department of Chemistry for providing organic chemistry laboratories to complete the requirements for reactions.

References

- 1. Schiff, H. (1864). Sur quelques dérivés phéniques des aldéhydes. Annali Di Chimica, 131, 118.
- Kahlal, S., Saillard, J.-Y., Hamon, J.-R., Manzur, C., & Carrillo, D. (1998). Molecular orbital analysis of the metal-hydrazide (2-) bonding in coordination chemistry. *Journal of the Chemical Society, Dalton Transactions*(7), 1229-1240.
- 3. Layer, R. W. (1963). The Chemistry of Imines. *Chemical reviews*, 63(5), 489-510.
- 4. Patai, S. (1970). Chemistry of the carbon-nitrogen double bond.
- 5. A.K. Sen-Gupta and K. Gajela; J. Ind. Chem. Soc., LVIIII, 690 (1981).
- Pollack, R. M., Kayser, R. H., & Damewood Jr, J. R. (1977). The effect of solvent on intramolecular general base catalysis in the hydrolysis of. alpha.,. beta.-unsaturated Schiff bases. *Journal of the American Chemical Society*, 99(25), 8232-8237.
- 7. Aljamali, N. M. (2018). Reactions and Mechanisms. 1 Edt., IJMRA Publication.
- 8. Khalid Mohammad Daoud and Mohammad Nizar Ibrahim, "Mechanism of Organic Chemistry", Al-Mousel University, (247 252), (1999).
- 9. Chaturvedi, R. K., & Cordes, E. (1967). The Temperature Dependence of the Hydrolysis of a Series of Substituted Benzylidene-1, 1-dimethylethylamines. *Journal of the American Chemical Society*, 89(5), 1230-1234.
- 10. A.D.M. El- Dib, M.Sc. Thesis, Baghdad Univ., (1996).
- 11. Reeves, R. (1962). p-Trimethylammoniumbenzylidene-p'-hydroxyaniline Chloride in Aqueous Solution from p H 1 to 11.5. *Journal of the American Chemical Society*, 84(17), 3332-3337.
- 12. Rinartha, K., & Suryasa, W. (2017). Comparative study for better result on query suggestion of article searching with MySQL pattern matching and Jaccard similarity. In 2017 5th International Conference on Cyber and IT Service Management (CITSM) (pp. 1-4). IEEE.
- 13. Rinartha, K., Suryasa, W., & Kartika, L. G. S. (2018). Comparative Analysis of String Similarity on Dynamic Query Suggestions. In 2018 Electrical Power, Electronics, Communications, Controls and Informatics Seminar (EECCIS) (pp. 399-404). IEEE.
- 14. Santoso, P., Adrianta, K. A., & Wiranatha, I. G. (2021). Phytochemical screening and in vivo test of dewandaru (Eugenia uniflora L) fruit extract on

6120

mice exposed to cigarette smoke. International Journal of Health & Medical Sciences, 4(2), 246-252. https://doi.org/10.31295/ijhms.v4n2.1722