

Figure 1. Chromatogram of CTD01

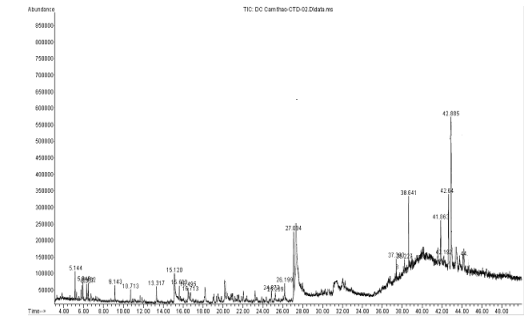


Figure 2. Chromatogram of CTD02

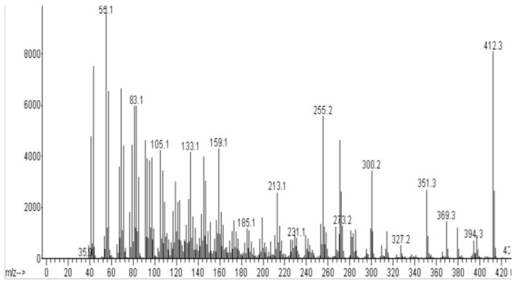


Figure 6. Mass Spectra of CTD02.2

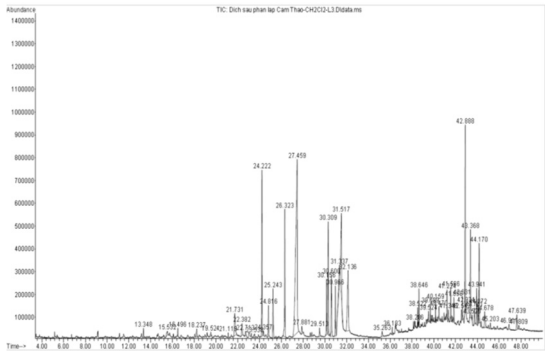


Figure 3. Chromatogram of CTD03

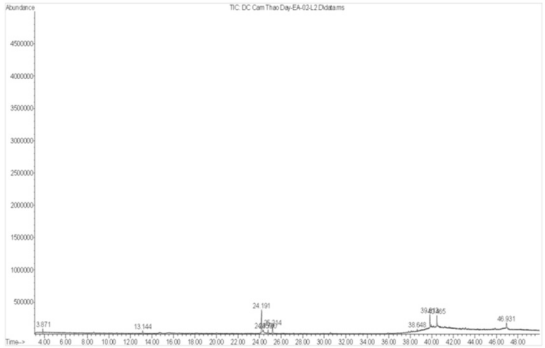


Figure 4. Chromatogram of CTD04

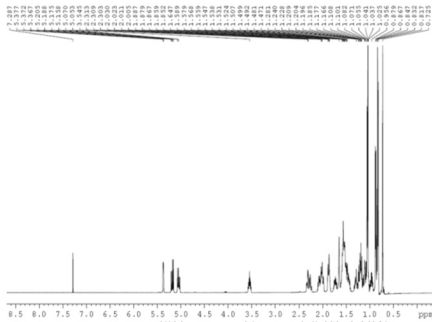


Figure 7. ¹H-NMR spectroscopy of CTD02.2

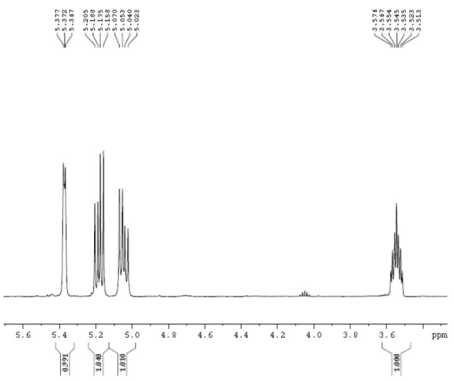


Figure 8. ¹H-NMR spectroscopy of CTD02.2 (δ 3-6ppm)

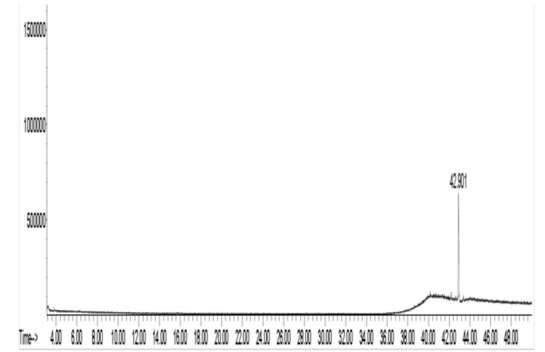


Figure 5. Chromatogram of CTD02.2

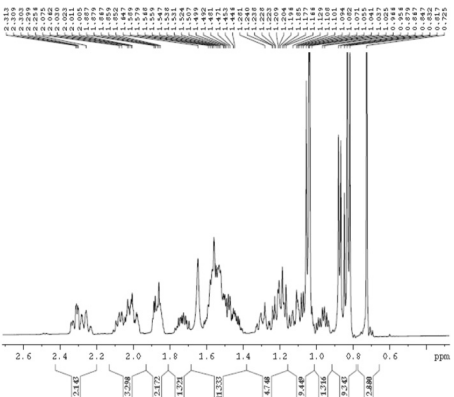


Figure 9. ¹H-NMR spectroscopy of CTD02.2 (δ 0.3 - 3 ppm)

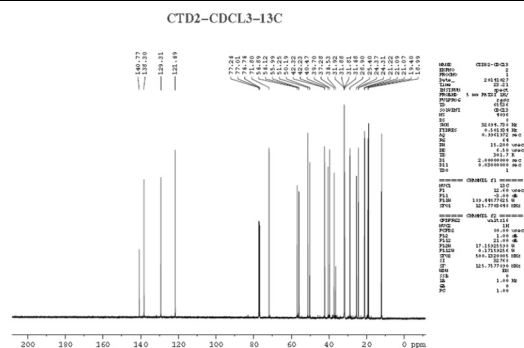


Figure 10. ^{13}C -NMR spectroscopy of CTD02.2

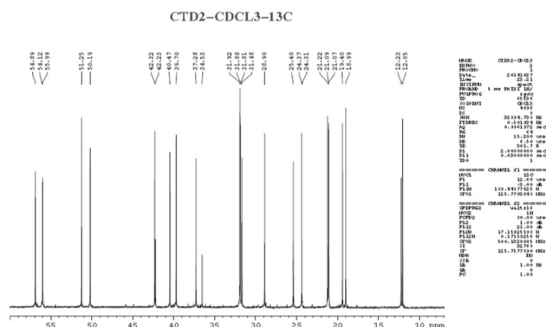


Figure 11. ^{13}C -NMR spectroscopy of CTD02.2 (δ 10-60 ppm)

3. Results and Discussion

The CTD02.2 is a white solid crystal.

MS of CTD02.2: m/z base peak 412.3(M^+ , $\text{C}_{29}\text{H}_{48}\text{O}$), 394.3, 369.3, 351.3, 327.2, 300.2, 273.2, 255.2, 231.1, 213.1, 199.4, 185.1, 159.1, 145.6, 133.1, 121.2, 105.1, 91.4, 83.1, 69.7, 55.1, 41.5.

The IR(KBr) absorption spectrum shows absorption peaks at 3422.15cm^{-1} , 2937.44cm^{-1} and 2867.32cm^{-1} , 1654.04cm^{-1} , 1636.75cm^{-1} , 1382.12cm^{-1} , 1192.60cm^{-1} , 1051.86cm^{-1} , 970.45cm^{-1} , 799.83cm^{-1} , 591.93cm^{-1} and 419.35cm^{-1} .

^1H -NMR (500 MHz, CDCl_3) of CTD02.2: ^1H -NMR has given signals at δ 7.287, 5.377, 5.372, 5.367, 5.205, 5.188, 5.175, 5.158, 5.070, 5.053, 3.545, 2.313, 2.309, 2.303, 2.030, 2.023, 2.011, 2.005, 1.887, 1.879, 1.867, 1.859, 1.852, 1.647, 1.589, 1.579, 1.568, 1.559, 1.547, 1.538, 1.531, 1.524, 1.507, 1.499, 1.492, 1.481, 1.471, 1.281, 1.240, 1.228, 1.209, 1.204, 1.196, 1.185, 1.177, 1.166, 1.108, 1.101, 1.082, 1.071, 1.055, 1.042, 1.037, 1.025, 0.956, 0.879, 0.867, 0.847, 0.832, 0.817, 0.725.

^{13}C -NMR (125 MHz, CDCl_3) of CTD02.2: ^{13}C -NMR has given signals at 37.28 (C-1), 31.88 (C-2), 71.80(C-3), 42.23(C-4), 140.77(C-5), 121.69(C-6), 31.92(C-7), 31.88(C-8), 50.19(C-9), 36.53(C-10), 21.06(C-11), 39.70(C-12), 42.23(C-13), 56.89(C-14), 24.37(C-15), 28.90(C-16), 55.99(C-17), 12.23(C-18), 21.07(C-19), 40.47(C-20), 21.22(C-21), 138.30(C-22), 129.31(C-23), 51.24(C-24), 31.88(C-25), 18.99(C-26), 19.40(C-27), 25.40(C-28), 12.05(C-29).

The MS spectrum shows a parent molecular ion $[\text{M}+\text{H}]^+$ peak at m/z 412.3 which correspond to the molecular formula $\text{C}_{29}\text{H}_{48}\text{O}$.

The IR signal absorption band observed at 3422.15cm^{-1}

1 is characteristic of O-H stretching. Absorption at 2937.44cm^{-1} and 2867.32cm^{-1} is due to aliphatic C-H stretching. Other absorptions at $1636.75 - 1654.06\text{cm}^{-1}$ are because of C=C stretching, however this band is weak. Absorption at 1458.46cm^{-1} is a bending frequency for cyclic $(\text{CH}_2)_n$. Absorption at 1382.12cm^{-1} is attributable to OH deforming absorption. The absorption frequency at 1051.86cm^{-1} signifies cycloalkane. These absorption frequencies resemble the absorption frequencies observed for Stigmasterol.

Similarly, from ^1H -NMR data of CTD02.2, it is seen that carbinyl proton appears at δ 3.51 (1H, m, H-3). Three vinylic protons appears at δ 5.37 (1H, d, $J = 5\text{Hz}$, H-6), δ 5.02 (1H, dd, $J = 15.0\text{Hz}$ and 8.5Hz , H-22), δ 5.16 (1H, dd, $J = 15.0\text{Hz}$ and 8.5Hz , H-23). Six methyl protons also appears at δ 0.73 (3H, s, H-18), δ 1.06 (3H, s, H-19); δ 0.81 (3H, d, $J = 7.5\text{Hz}$, H-27), δ 0.86 (3H, d, $J = 7.5\text{Hz}$, H-26) và δ 0.82 (3H, d, $J = 7.5\text{Hz}$, H-21); $\delta = 0.84$ (3H, t, H-29) [3].

The ^{13}C -NMR of CTD02.2 shows a total of 29 carbons. Signals δ 140.77 ppm and 121.69 ppm are assignable to the double bond at C_5 and C_6 [4]. The alkene carbons appear at 138.30 ppm (C_{22}) and 129.31 ppm (C_{23}) [5]. The δ value observed at 71.80 ppm is due to C_3 hydroxyl group[6]. The value at δ 12.23 ppm and 19.40 ppm corresponds to angular carbon atoms (C_{18} and C_{19}). Spectra shows 29 carbon signals including six methyls, nine methylenes, eleven methanes and three quaternary carbons.

Table 1. ^1H -NMR and ^{13}C -NMR data of compound CTD02.2 and Stigmasterol.

STT	Compound CTD 02.2		Stigmasterol [3]	
C	δ_{C}	δ_{H} (J=Hz)	δ_{C}	δ_{H} (J=Hz)
1	37.28		37.28	
2	31.88		31.67	
3	71.80	3.50 (1H, m, H-3)	71.78	3.49 (1H, m, H-3)
4	42.23		42.32	
5	140.77		140.76	
6	121.69	5.37 (1H, d, $J = 5,2\text{ Hz}$, H-6)	121.66	5.33 (1H, d, $J = 5,2\text{ Hz}$, H-6)
7	31.92		31.92	
8	31.88		31.90	
9	50.19		50.20	
10	36.53		36.52	
11	21.06		21.08	
12	39.70		39.70	
13	42.23		42.23	
14	56.89		56.88	
15	24.37		24.36	
16	28.90		28.89	

17	55.99		55.99	
18	12.23	0.73 (3H, s, H-18)	12.22	0.69 (3H, s, H-18)
19	21.07	1.06 (3H, s, H-19)	21.06	1.01 (3H, s, H-19)
20	40.47		40.50	
21	21.22	0.96 (3H, d, J = 7.5Hz, H- 21)	21.22	0.92 (3H, d, J=6.5Hz, H- 21)
22	138.30	5.05 (1H, dd, J = 8.5Hz and 15.0Hz, H-22)	138.28	5.03 (1H, dd, J=8.5Hz, 15.5Hz, H-22)
23	129.31	$\delta = 5.16$ (dd, J = 8.5Hz and 15.0 Hz, H- 23)	129.30	5.11 (1H, J=15.5Hz, 8.5Hz, H-23)
24	51.24		51.25	
25	31.88		31.87	
26	18.99	0.85 (3H, d, J = 7.5Hz, H- 26)	18.99	0.84 (3H, d, J=6.5Hz, H- 26)
27	19.40	0.81 (3H, d, J= 7.5Hz, H-27)	19.39	0.81 (3H, d, J=6.5Hz, H- 27)
28	25.4		25.39	
29	12.05	0.82 (3H, d, J= 7.5Hz, H-29)	12.05	

4. Conclusion

We have isolated the compound CTD02.2 from the stem and foliage of red form of *Abrus precatorius* Linn collected from the Dai Loc District, Quang Nam Province, Vietnam. The isolated compound is purified by silica gel column chromatography. From the above IR, $^1\text{H-NMR}$, $^{13}\text{C-NMR}$ and MS spectral data with the comparison made so far, it is concluded that sample CTD02.2 is Stigmasterol (Figure 5).

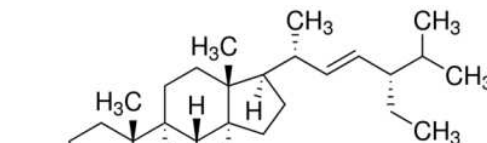


Figure 5. Chemical structure of CTD02.2 (Stigmasterol)

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