

DEVELOPMENT OF TECHNOLOGY FOR PRODUCING NP-FERTILIZER BASED ON LOCAL RAW MATERIALS

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ANNOTATION:

The article presents the results of nitric and sulfuric acid decomposition of unfortified phosphate rock. In order to determine the degree of decomposition of phosphate raw materials, depending on the concentration and rate of sulfuric acid, the calculated amount of phosphorite was treated with an acid with thorough stirring for 20 min. The resulting product, after cooling and after drying at a temperature of 105-110 °C, was subjected to chemical analysis. Fertilizer production by decomposition of low-grade phosphorites at an incomplete rate of concentrated sulfuric acid and weak nitric acids occurs in a solid-phase mode and differs from classical methods in that when the interaction of the components is completed, the resulting mass is constantly fragmented into small aggregated particles, which, in turn, are continuously enveloped with excess phosphate raw materials.

Keywords: phosphorite, sulfuric acid, superphosphate, degree of decomposition, incomplete rate, acid concentration.

Along with the harvest, there is an intensive removal from the soil of all the nutrients of mineral fertilizers. The deficiency of these elements in soils must be replenished by the application of fertilizers, taking into account the utilization rate of various forms of active substances. However, the utilization rate of fertilizers applied to the soil by plants is still low. Therefore, increasing the efficiency of using mineral fertilizers through the production of higher quality species is an important task. From the above, it follows that it is necessary to create new types of phosphorus and potassium-containing fertilizers and to increase their production. The President of Uzbekistan and the Government of the Republic pay great attention to the issues of building up these industries.

The range of mineral fertilizers produced by the domestic fertilizer industry in terms of the ratio of nutrients does not fully meet the

requirements of agrochemical science for the organization of highly efficient cultivation of agricultural crops. Therefore, an important problem remains the development and implementation of fundamentally new, more economical (unconventional) technologies for processing low-grade local phosphorite ores, including for balance raw materials, into highly effective complex fertilizers with the best technical and economic indicators.

Previous studies on the decomposition of high-carbonate phosphorites of the Central Kyzyl Kum with concentrated sulfuric acid make it possible to develop intensive technologies for obtaining complex phosphorus-containing fertilizers.

The essence of the process of obtaining phosphorus-containing complex fertilizers such as superphosphate is expressed by the reaction of interaction of fluorapatite ($\text{Ca}_5(\text{PO}_4)_3\text{F}$) - the main component of phosphate raw materials at an incomplete rate with sulfuric and nitric acids.

To determine the optimal conditions for the process of nitric and sulfuric acid decomposition of unfortified phosphate rock, washed phosphoconcentrate and poor phosphorites, sulfuric acid with a

concentration of 75, 80 and 93 % and nitric acid with a concentration of 57 – 59 % were used. The rate of sulfuric and nitric acids was calculated for the decomposition of phosphate and carbonate minerals of phosphorite to form monocalcium phosphate, calcium sulfate, and calcium nitrate.

In order to determine the degree of decomposition of phosphate raw materials, depending on the concentration and rate of sulfuric acid, the calculated amount of phosphorite was treated with an acid with thorough stirring for 20 min. The resulting product, after cooling and after drying at a temperature of 105-110 °C, was subjected to chemical analysis. The results are shown in Tables 1-4.

Fertilizer production by decomposition of low-grade phosphorites at an incomplete rate of concentrated sulfuric acid and weak nitric acids occurs in a solid-phase mode and differs from classical methods in that when the interaction of the components is completed, the resulting mass is constantly fragmented into small aggregated particles, which, in turn, are continuously enveloped with excess phosphate raw materials.

Table 1. Influence of the norm of sulfuric (H_2SO_4 -93 %) and nitric (HNO_3 -59 %) acids on the degree of decomposition of unfortified phosphate rock

Acid rate,%		P ₂ O ₅ , %		CaO, %	N, %	K _p , %	H ₂ O, %
H ₂ SO ₄	HNO ₃	general.	digestible				
Before drying							
60	20	11,19	9,76	25,11	1,95	87,22	6,40
50	30	10,20	8,98	22,98	2,60	88,04	8,40
40	40	9,88	8,73	22,17	3,46	88,36	10,81
30	50	9,43	8,36	21,17	4,11	88,65	12,79
20	60	8,93	7,95	20,03	4,70	89,03	14,71
After drying							
60	20	11,78	10,38	26,43	2,05	88,12	1,47
50	30	10,79	9,69	24,61	2,80	88,33	1,51
40	40	10,89	9,60	24,36	3,80	88,40	1,99
30	50	10,60	9,38	23,79	4,65	88,49	2,21
20	60	10,26	9,10	23,02	5,40	88,69	1,97

Table 2 Influence of the norm of sulfuric (H₂SO₄-80 %) and nitric (HNO₃-59 %) acids on the degree of decomposition of unfortified phosphate rock

Acid rate,%		P ₂ O ₅ , %		CaO, %	N, %	K _p , %	H ₂ O, %
H ₂ SO ₄	HNO ₃	general.	digestible				
Before drying							
60	20	10,94	9,52	24,50	1,91	87,02	7,01
50	30	10,16	9,89	22,79	2,67	88,19	9,37
40	40	9,85	8,46	21,49	3,36	88,31	14,52
30	50	9,20	8,16	20,64	4,03	88,70	17,92
20	60	8,81	7,83	19,76	4,63	88,88	18,52
After drying							
60	20	11,52	10,06	25,79	2,01	87,33	1,06
50	30	10,92	9,64	24,51	2,87	88,28	1,47
40	40	10,53	9,32	23,62	3,63	88,51	1,67
30	50	10,22	9,08	22,93	4,48	88,85	2,13
20	60	10,13	9,01	22,71	5,32	88,94	1,75
60	20	11,24	9,89	25,25	1,97	87,99	2,01
50	30	10,63	9,40	28,89	2,79	88,43	1,27
40	40	10,64	9,43	23,89	3,74	88,63	1,78
30	50	10,34	9,16	23,22	4,54	88,59	1,83
20	60	10,08	8,94	22,66	5,31	88,69	1,67

The results of the chemical analysis of the obtained complex fertilizer show that with an increase in the norm of nitric acid from 20% to 60%, the nitrogen content increases from 1.97% to 5.31% and the water-soluble form of CaO from 3.94% to 10.62% in the form of four aqueous calcium nitrate.

Complex fertilizer before drying, obtained by decomposition of unenriched phosphate flour with a mixture of acids at their ratio H₂SO₄: HNO₃ = 60: 20 and 50:30, has satisfactory commercial properties. Further

increase in the rate of nitric acid leads to a smearing mass.

It was found that the use of sulfuric acid with a concentration of 75 and 80% leads to an increase in the moisture content in the composition of the complex fertilizer.

The main components of a complex fertilizer obtained by nitric-sulfuric acid decomposition of unfortified phosphate rock are mono and dicalcium phosphate, calcium sulfate and nitrate.

Table 3 Influence of the norm of sulfuric (H₂SO₄-93%) and nitric (HNO₃-57%) acids on the degree of decomposition of poor phosphorites

Acid rate,%		P ₂ O ₅ , %		CaO, %	N, %	H ₂ O, %	CO ₂ , %	K _p , %
H ₂ SO ₄	HNO ₃	general.	digestible					
Before drying								
60	20	10,20	8,95	26,74	2,14	8,19	1,78	87,74
50	30	10,10	8,81	26,26	3,16	9,34	1,67	87,00
40	40	9,65	8,49	25,05	4,03	12,32	1,50	87,98
30	50	9,26	8,17	24,08	4,81	15,53	0,83	88,23
20	60	9,94	8,82	25,80	6,22	20,75	0,90	88,73
After drying								
60	20	11,08	9,84	28,35	2,31	1,40	1,80	88,80
50	30	11,10	9,76	28,79	3,43	1,40	1,69	87,93
40	40	10,83	9,55	28,12	4,52	1,44	1,58	88,18
30	50	10,77	9,57	27,96	5,62	1,71	0,95	88,85
20	60	12,16	10,81	31,56	7,62	2,31	0,97	88,90

Table 4. Influence of the norm of sulfuric (H₂SO₄-75 %) and nitric (HNO₃-57 %) acids on the degree of decomposition of washed fosconcentrate

Acid rate,%		P ₂ O ₅ , %		CaO, %	N, %	H ₂ O, %	CO ₂ , %	K _p , %
H ₂ SO ₄	HNO ₃	general.	digestible					
Before drying								
60	20	10,31	9,06	20,32	0,97	5,52	1,28	87,87
50	30	9,72	8,56	19,16	1,31	7,80	1,14	88,06
40	40	9,19	8,10	18,20	1,65	9,84	1,02	88,14
30	50	8,72	7,71	17,20	1,96	11,67	0,9	88,42
20	60	8,31	7,36	16,39	2,24	13,36	0,81	88,57
After drying								
60	20	10,74	9,45	21,17	0,98	1,58	1,33	87,99
50	30	10,34	9,12	20,38	1,40	1,91	1,21	88,20
40	40	10,00	8,85	19,70	1,80	2,00	1,11	88,50
30	50	9,69	8,60	19,10	2,18	1,86	1,01	88,75
20	60	9,44	8,39	18,63	2,55	1,55	0,91	88,88

The decomposition of washed phosphoconcentrate and poor phosphorites by a mixture of nitric and sulfuric acids practically does not differ from that of unfortified phosphate rock.

It was found that with an increase in the rate of sulfuric acid, the content of calcium sulfate increases, and with an increase in the rate of nitric acid, a highly hygroscopic calcium nitrate in the composition of a complex fertilizer.

The interaction of unfortified phosphate rock with a mixture of sulfuric and nitric acids proceeds very easily and is practically completed in 5-7 minutes. The process is exothermic, the temperature, depending on the rate of sulfuric acid, rises to 35-80 ° C.

The main components of a complex fertilizer obtained by nitric-sulfuric acid decomposition of various types of phosphate raw materials are four aqueous calcium nitrate, two aqueous calcium sulfate, calcium phosphates, and activated phosphorite. The content of calcium nitrate increases with an increase in the norm of nitric acid.

The commercial properties of nitrogen-phosphorus complex fertilizers have been studied.

On the basis of the results obtained, a schematic diagram was developed for obtaining a complex nitrogen-phosphorus fertilizer by intensive decomposition of high-carbonate phosphorites with a mixture of nitric and sulfuric acids.

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